| | | STUDY MODULE D | ESCRIP | TION FORM | | | | |
|---|--|--|---|--|-----------------------------|---|--|--|
| Name of the module/subject New Methods of Organic Compounds Synthesis | | | | | Code 1010702211010720081 | | | |
| Field of | ^{study} mical Technolog | у | | of study al academic, practical IK) |) | Year /Semester | | |
| | path/specialty | anic Technology | Subjec | t offered in: Polish | | Course (compulsory, elective) obligatory | | |
| Cycle o | | | Form of stud | dy (full-time,part-time) | | obligatory | | |
| Second-cycle studies | | | | full-time | | | | |
| No. of h | iours | | 1 | | | No. of credits | | |
| Lectu | re: 2 Classes | s: 1 Laboratory: 3 | Projec | t/seminars: | - | 5 | | |
| Status | of the course in the study | program (Basic, major, other) | (universit | ty-wide, from another | field) | | | |
| | | (brak) | | | (bra | ak) | | |
| Educati | on areas and fields of sci | ence and art | | | | ECTS distribution (number and %) | | |
| techi | nical sciences | | | | | 5 100% | | |
| | Technical scie | ences | | | | 5 100% | | |
| | | | | | | | | |
| Responsible for subject / lecturer: dr hab. inż. Aleksandra Borowiak-Resterna email: aleksandra.borowiak-resterna@put.poznan.pl tel. 616653689 Faculty of Chemical Technology | | | | | | | | |
| | Piotrowo 3 60-965 Poz | | | | | | | |
| | | s of knowledge, skills an | d social o | competencies: | | | | |
| 1 | Knowledge | chemistry obtained during studie characteristic, basic reactions of | neoretically founded knowledge of general and organic es in chemical technology first degree, knows the the most important groups of organic compounds. | | | | | |
| | | 2. The student has a basic knowledge of information technology. 1. The student is able to solve the basic problem tasks of organic chemistry based on his | | | | | | |
| 2 | Skills knowledge. 2. The student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability to obtain information from the identity of the student has the ability of the student has the student has the ability of the student has the ability of the student has the student has the ability of the student has the student has the ability of the student has the student has the ability of the student has the ability of the student has the student has the ability of the student has the student has the ability of the student has the ability of the student has the ability of the student has the student has the student has the student has the ability of the student has the stud | | | | | ified sources | | |
| | Social | | | o supplement his knowledge and its continuous updating | | | | |
| 3 | competencies | 2. The student knows the basic | | | | | | |
| ٨٩٩١ | • | ectives of the course: | p | | | | | |
| 1. Prov | vide students with exte | ensive and solid knowledge of orgation of obtaining the selected group of the selected g | | | netic i | methods, how to plane and | | |
| 2. Dev | elop students' ability to e recipe preparative of | o solve basic problems in the one- f selected organic compound base | -step and m | ulti-step syntheses | | | | |
| 3. Mas | tering the students' ab | ility to use molecular modeling to plex organic molecules as well as | | | | | | |
| 4. Develop students' awareness of the responsibility for their future decisions relevant to the chemical engineer work. | | | | | | | | |
| | Study outco | mes and reference to the | educatio | onal results for | r a f | ield of study | | |
| Knov | vledge: | | | | | | | |
| | | ded, solid knowledge of modern m taining the selected groups of org | | | netho | ds of planning and selection | | |
| | 2. The student has in-depth knowledge of new, versatile and efficient chemical reactions that allow to reduce or even exclude some of the issues related to environmental protection during the synthesis on an industrial scale [K_W08] | | | | | | | |
| | | odern methods of study of the stru n of raw materials and products of | | | als ba | sed on molecular modeling, | | |
| Skills | s: | | | | | | | |

1. The student can reliable select source of chemical and environmental information, critically evaluate the information obtained from the literature and electronic databases and carry out their analysis and draw conclusions. - [K_U01]

2. The student based on source material (also in English) and his own research can make a professional presentation of studied scientific issue. - [K_U06]

3. The student is able to use professional software, using them for the design of chemical processes and intermolecular interactions. - [K_U07]

4. The student knows and respects the principles of work-related health and safety in the chemical laboratory. - [K_U18]

Social competencies:

1. The student understands the need for continuous professional development. - [K_K01]

2. The student understands the importance of the responsibility that rests on all members of the team performing the task assigned; is aware of the need to respect the principles of teamwork. - [K_K04]

Assessment methods of study outcomes

Lectures - Assessment of knowledge and skills on the basis of the written exam (6 problem tasks based on course description of lectures).

Classes - Assessment prepared by the student (based on reference literature) presentation discusses the selected type of modern chemical reactions and to assess the knowledge he has acquired on the basis of test summary. Laboratory:

1) Synthesis of the product - oral answer before the preparation, analyzing how to execute of the planned synthesis (based on the research literature); evaluation of the practical implementation of the synthesis of the desired product; evaluation report containing an analysis of research literature and discuss the progress and result of the synthesis of an organic compound.

2) Molecular modeling - final test.

Course description

Lectures:

Factors influencing the reactivity of molecules and chemical reactions direction: types of interactions between molecules, acidity and basicity of organic compounds. electrophiles and nucleophiles. hard and soft acids and bases - sort by Pearson. Hammett equation, the role of the catalyst and solvent. Classification of solvents. Empirical parameters of solvent polarity. Phase transfer catalysis. Chemo-, regio- and stereoselective reactions. Stereospecific reactions. Selectivity and mechanisms of organic reactions: direct and conjugate addition (?1,2-addition? and ?1,4-addition?), elimination, electrophilic and nucleophilic substitution, reactions of oxidation and reduction. Protection of functional groups. Retrosynthetic analysis. Selective formation of carbon-carbon and carbon-heteroatom bonds, for example a metathesis reaction of alkenes, aldol condensation, Michael addition, Robinson annulation, coupling reactions with the participation of organometallic catalysts, pericyclic reactions. The use of organosilicone compounds, and organo-boron(III) compounds in organic synthesis. Examples of multi-step syntheses of organic compounds present in the environment (total syntheses).

Learning to use the primary literature. The discussion of the subject areas, with the active participation of students, some types of modern chemical reactions used in the multi-stage syntheses, among other compounds present in the environment. Laboratory:

1) Carrying out research literature on methods of obtaining the selected organic compound. Synthesis of the preparation using professional equipment and advanced purified techniques. Before making the preparation - a reminder and update industrial safety rules in force in the organic chemistry lab.

2) The introduction of the basic principles of molecular modeling - spatial manipulation of molecules with certain models of structural parameters in two and three dimensions, the basic techniques of molecular structures, modeling and measurement of structural parameters, building of multi-functional molecules, minimizing the energy of the molecule or molecules in the vacuum system.

Basic bibliography:

1. McMurry J., Chemia organiczna, PWN, Warszawa 2007.

2. Clayden J., Greeves N., Warren S., Wothers P., Chemia organiczna, tom I, II i III, WNT, Warszawa 2009.

3. Makosza M., Fedoryński M., Podstawy syntezy organicznej. Reakcje jonowe i rodnikowe, Oficyna Wydawnicza Politechniki Warszawskiej, Warszawa 2006.

4. Skarżewski J., Wprowadzenie do syntezy organicznej, PWN, Łódź 1999.

5. Buza D., Sas W., Szczeciński P., Chemia organiczna. Kurs podstawowy, Oficyna Wydawnicza Politechniki Warszawskiej, Warszawa 2006

Additional bibliography:

1. Willis C., Wills M, Synteza organiczna, Wydawnictwo Uniwersytetu Jagiellońskiego, Kraków 2004.

2. Smith M.B., March J., Advanced Organic Chemistry, Reaction, Mechanism and Structure, J.Wiley & Sons, New Jersey 2007.

3. Vogel A.I., Preparatyka organiczna, WNT, Warszawa 2006.

4. Przewodnik do nomenklatury związków organicznych, Polskie Towarzystwo Chemiczne, Warszawa 1994.

Result of average student's workload

| Activity | | Time (working hours) |
|--|--------|-------------------------|
| 1. Participation in lectures, exercises | | 45 |
| 2. Participation in the laboratory | 45 | |
| 3. Participation in consultations related to the preparation of present | 8 | |
| synthesis and preparation of the student for classes laboratory tests | 5 | |
| 4. Preparation of the presentation | 7 | |
| 5. The research literature and preparing the report after class laboration | 5 | |
| 6. Preparation for final test | 7 | |
| 7. Preparation for the test carried out on exercises | 17 | |
| 8. Preparation for the exam and the presence of the exam | | |
| Student's wo | rkload | |
| Source of workload | hours | ECTS |
| Total workload | 139 | 5 |
| Contact hours | 101 | 3 |
| Practical activities | 57 | 2 |